

Evaluating impacts of freshwater salinization syndrome on mobilization of nutrients and metals from stormwater best management practices

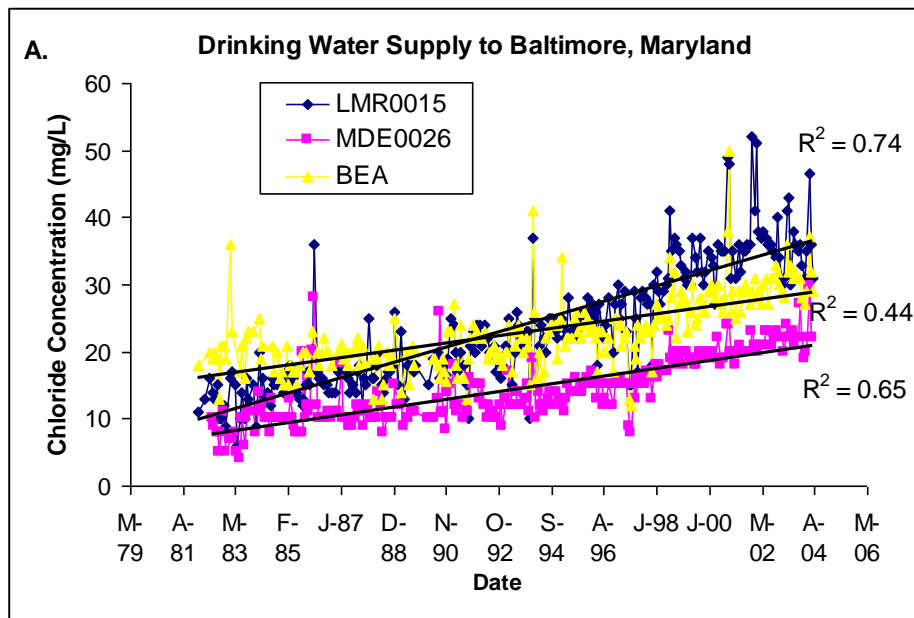
Research Team: Sujay Kaushal¹, Joseph Galella¹, Ruth Shatkay¹, Jenna Reimer¹, William Nguyen¹, Walter Boger¹, and Alexis Yaculak¹

Translation: Ken Mack²

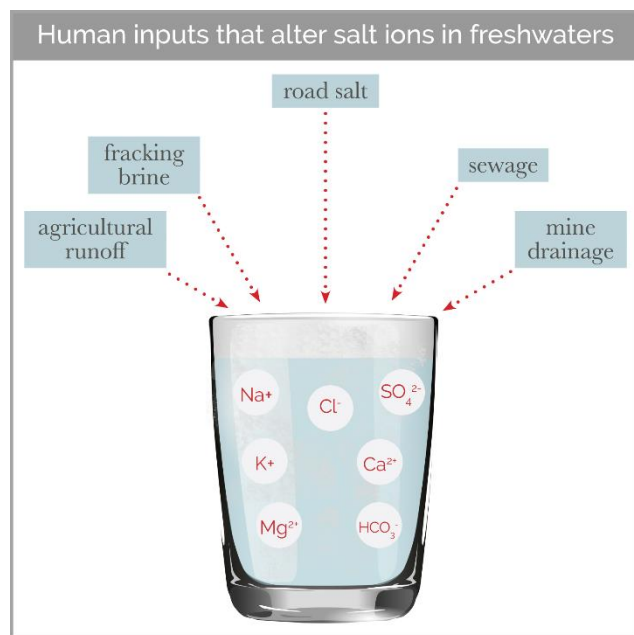
¹University of Maryland, Department of Geology & Earth System Science
Interdisciplinary Center

²Montgomery County, Department of Environmental Protection





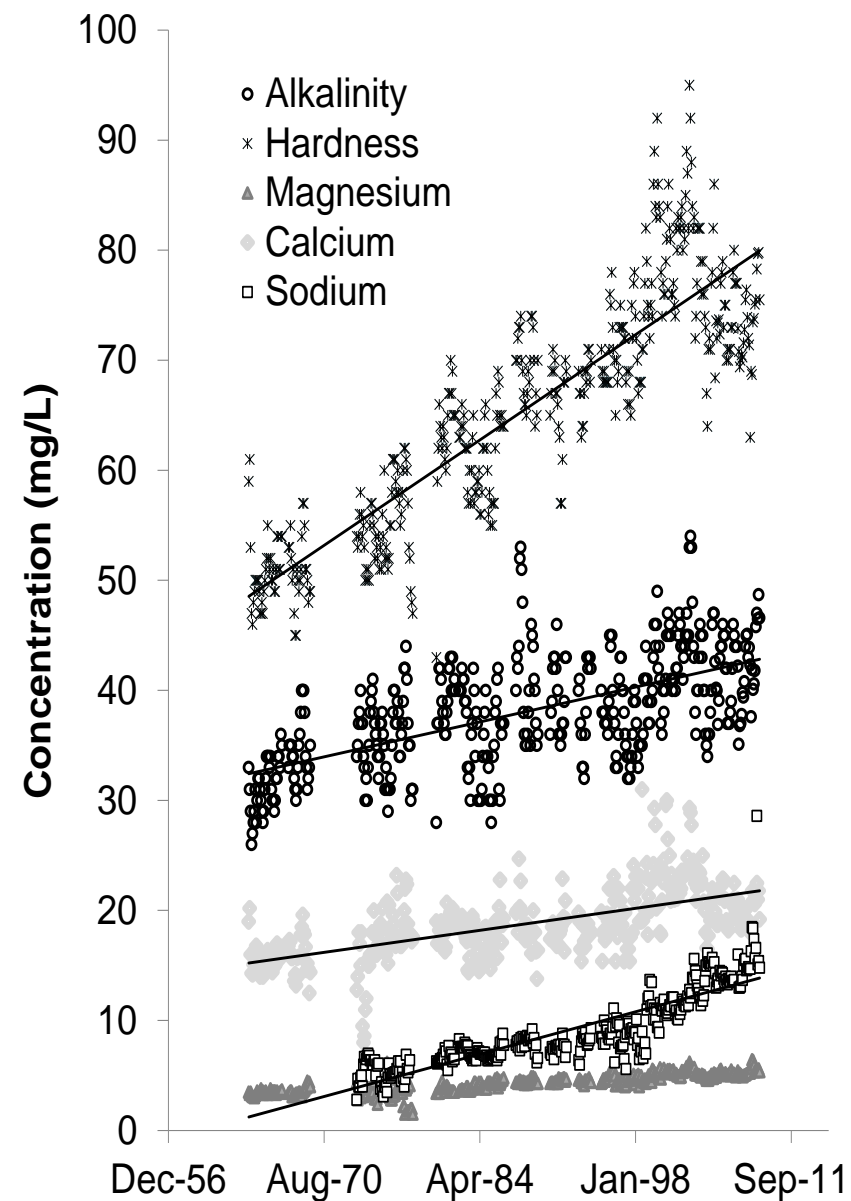
Kaushal et al. (2005) *PNAS*



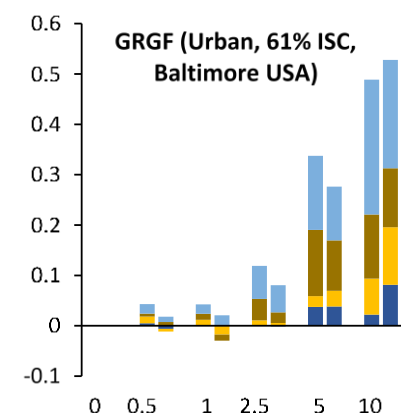
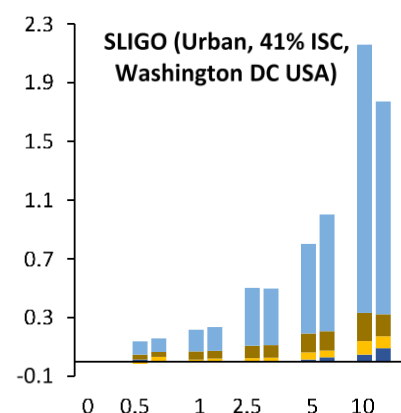
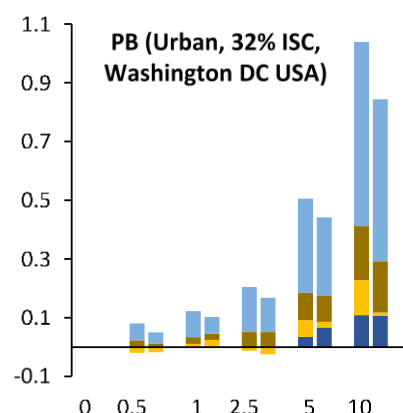
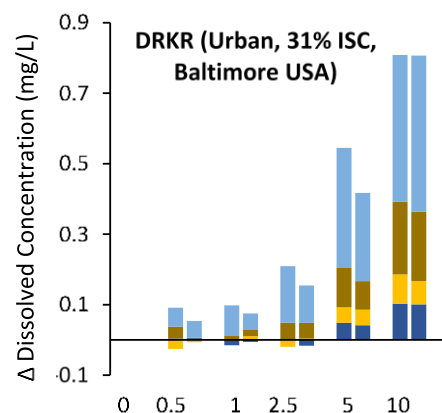
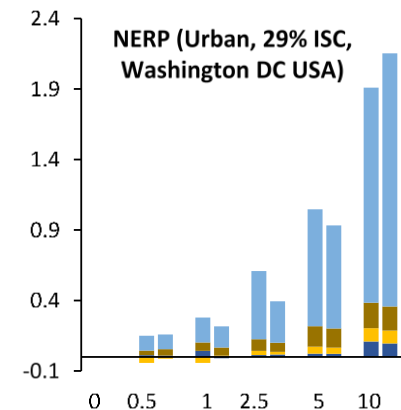
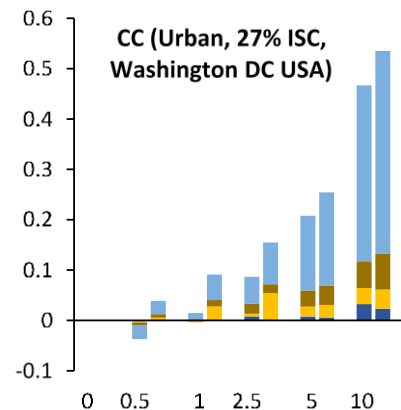
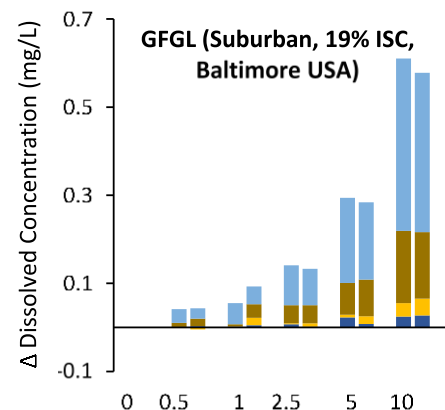
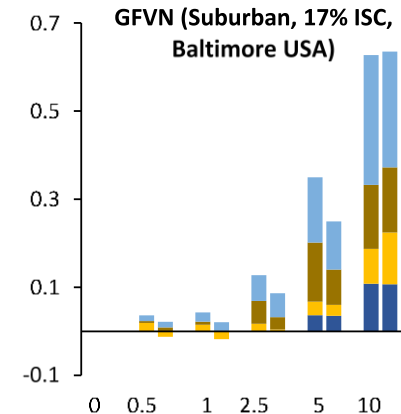
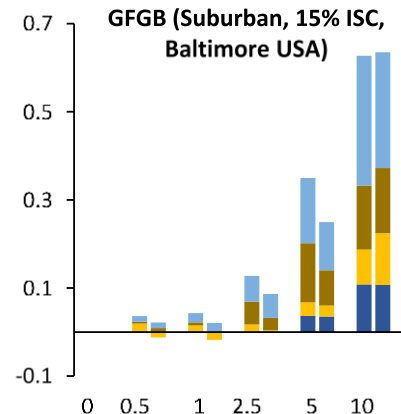
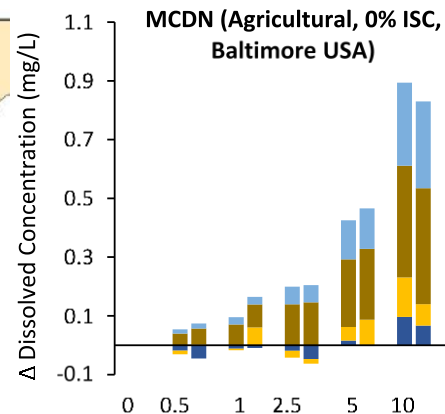
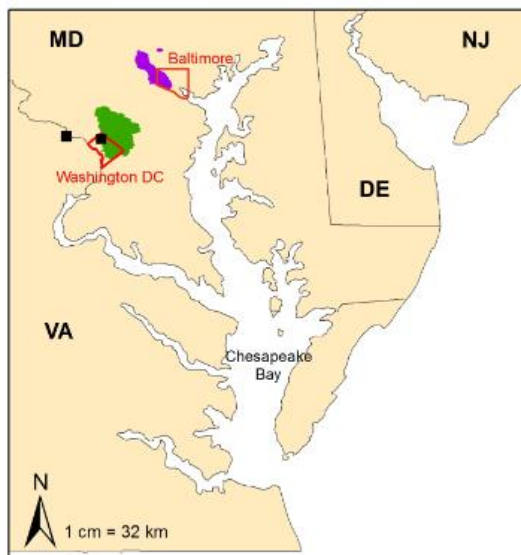
L. Quillen (2018) FSS Press Release

Freshwater Salinization Syndrome Impacts Maryland's Drinking Water

Thank You to Early Mentor: Bill Stack



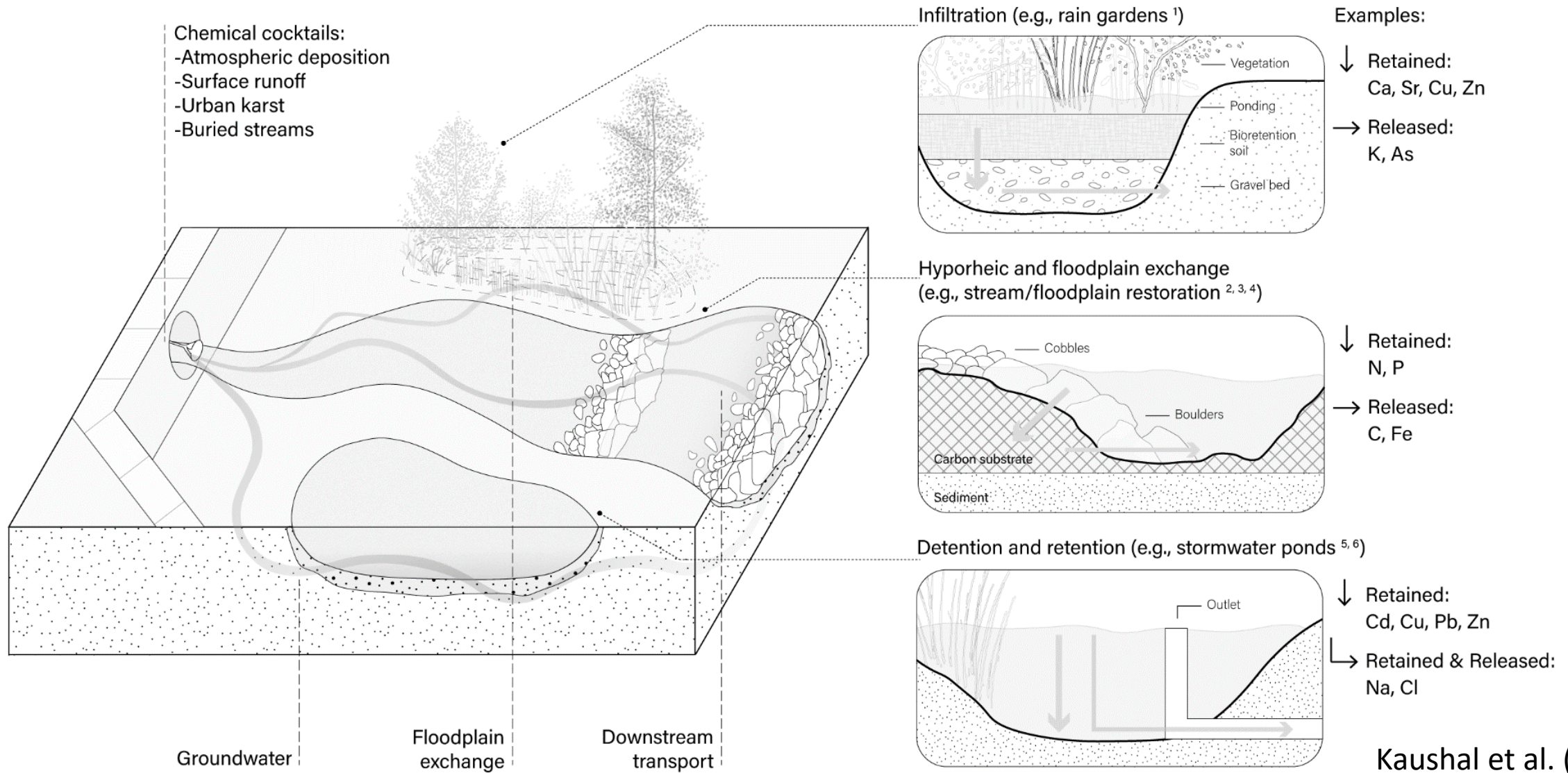
Kaushal et al. (2017) *Appl. Geochem*



Salinization Mobilizes Metals and Nutrients to Streamwater

Kaushal et al. (2019)
Philosophical Trans. Royal Society

Retention and release of chemical cocktails along stream and stormwater flowpaths

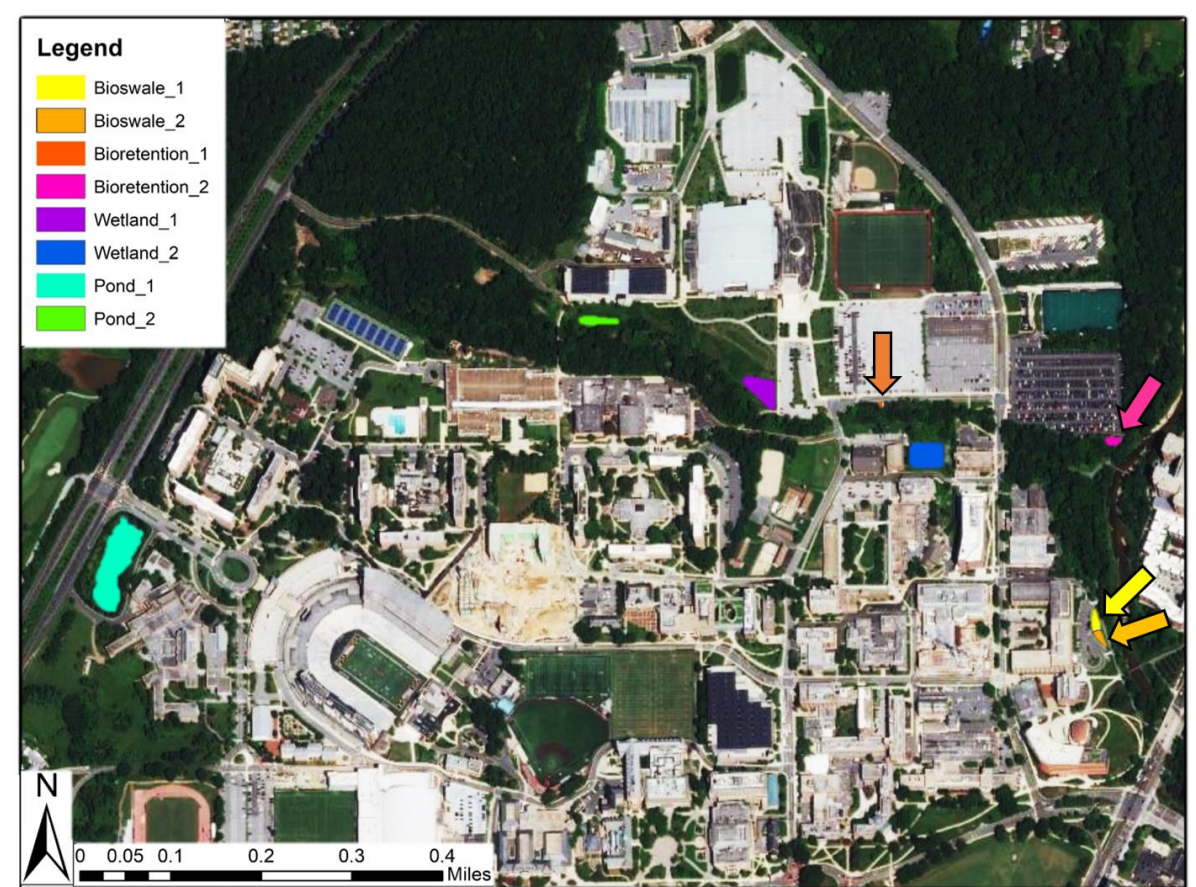


Key Questions

- What are critical thresholds in concentrations of different road salt ions (Na^+ , Ca^{2+} , Mg^{2+}) which can mobilize nutrients and metals to surface waters across varying stormwater BMPs?
- What are the concentrations and loads of different road salt ions and associated metals and nutrients in nearby stream outfalls before, during, and after deicing events

Study Sites:

- 4 different stormwater management features (with replicates)
- Sites range from 5 to 33 years in age and were almost always constructed concurrently with a building or large area of impervious surface coverage
- Stream restoration sites and regenerative stormwater conveyance (RSC) systems



Bioswale #1



Bioretention #1



Wetland #1



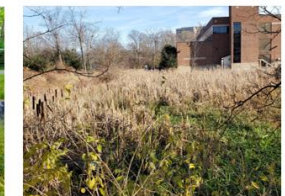
Pond #1



Bioswale #2



Bioretention #2



Wetland #2



Pond #2

Methods:

1. Collect water and sediment samples
2. Filter water through 0.7-micron GFF
3. Sieve sediment through a 4 mm sieve
4. Add salt (NaCl / CaCl_2 / MgCl_2)
5. Incubate water and sediment on shaker table for 24 hours

Joe Galella



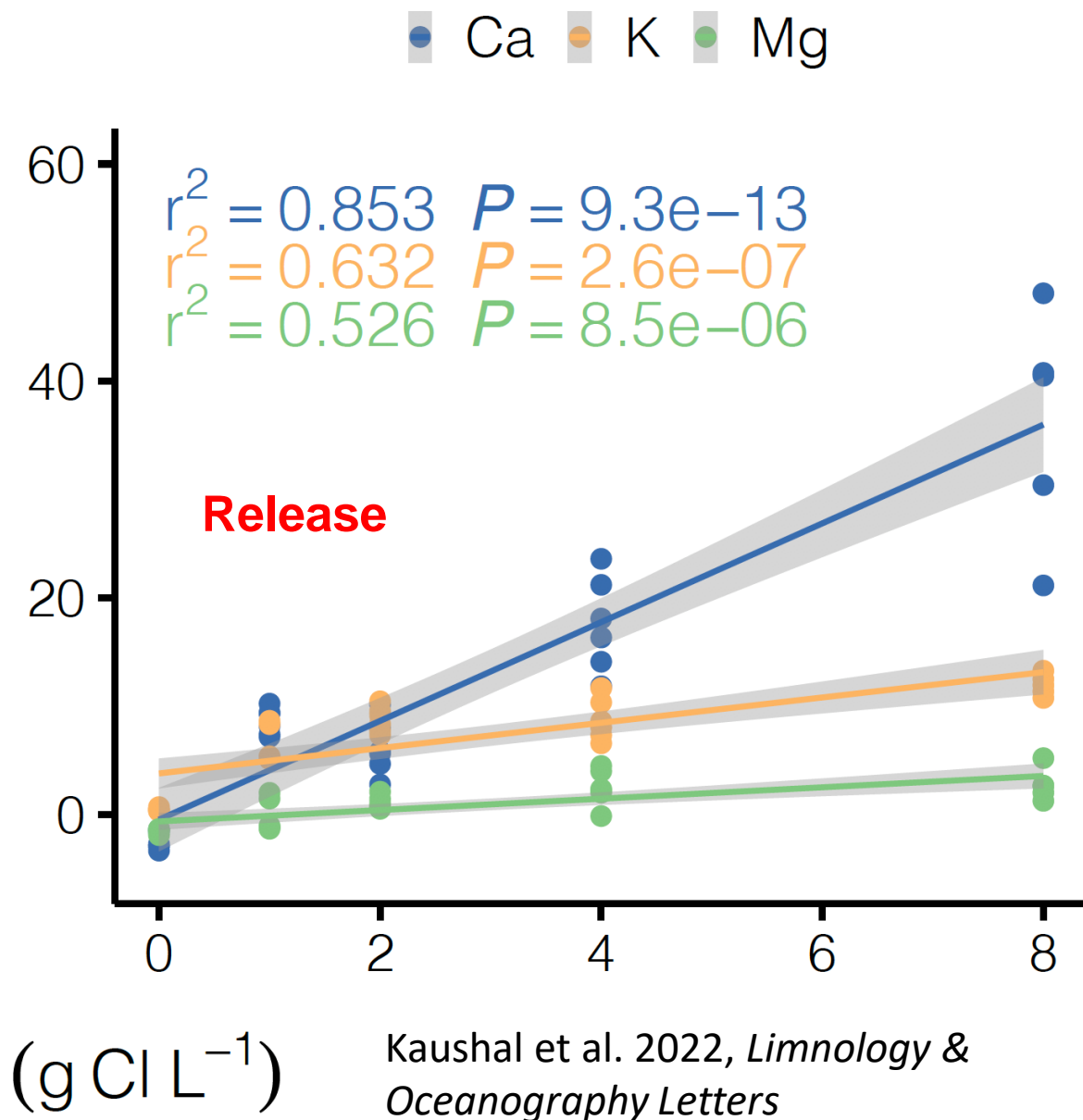
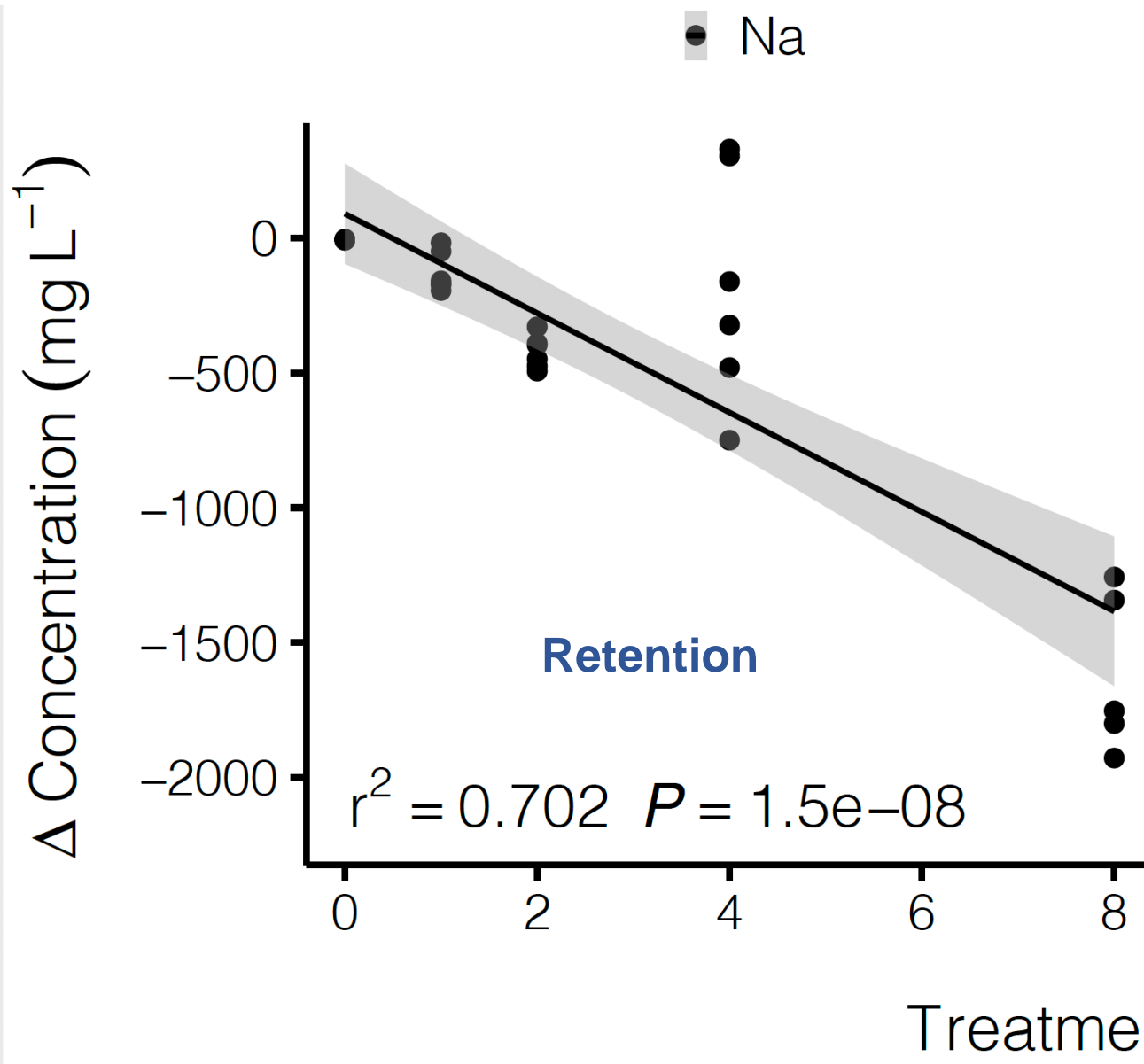
Hold the Salt: How Much Can Be Retained in Sediments?



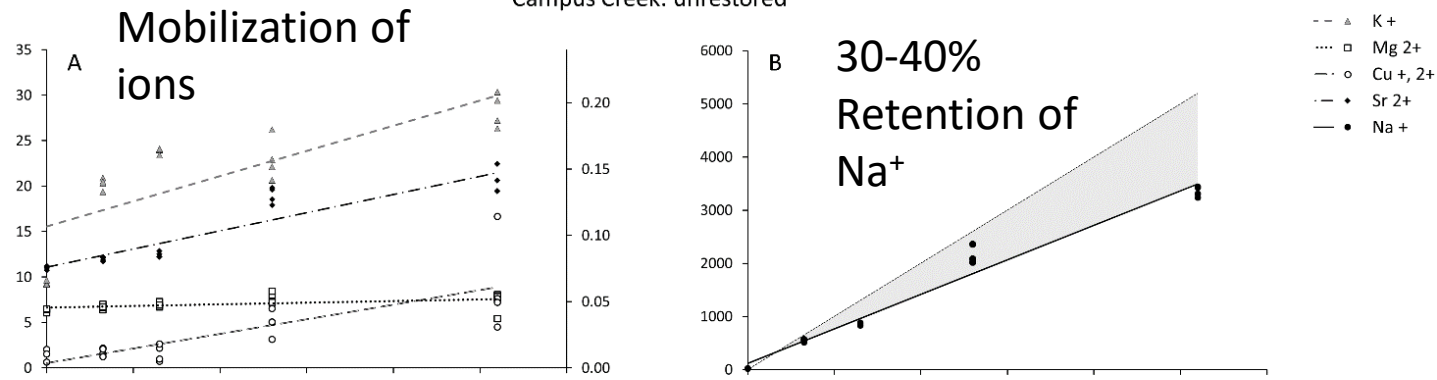
**Retention and Release of
Salts, Nutrients, and Metals in
Different Stormwater
Management Features**

Photo courtesy of
Kelsey Wood (2019)

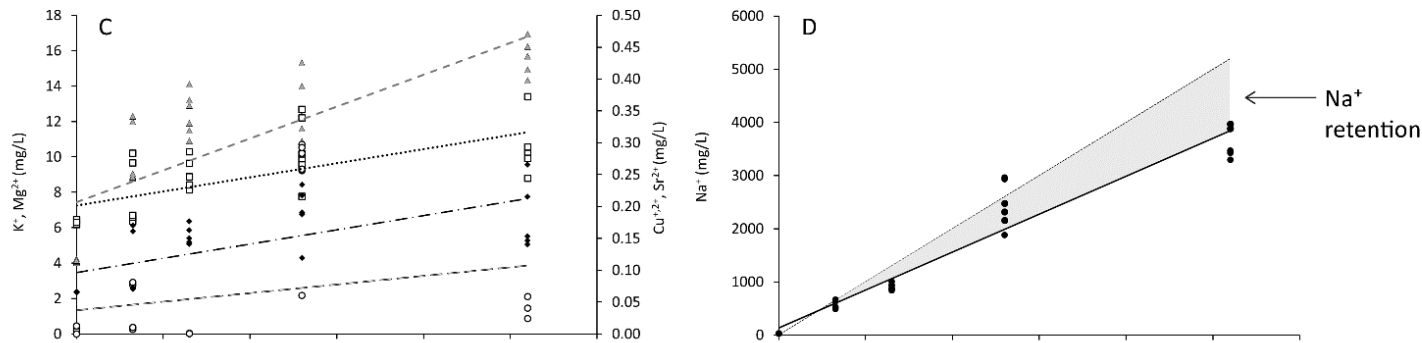
High Capacity for Sodium Retention in Restored Stream Floodplain Sediments



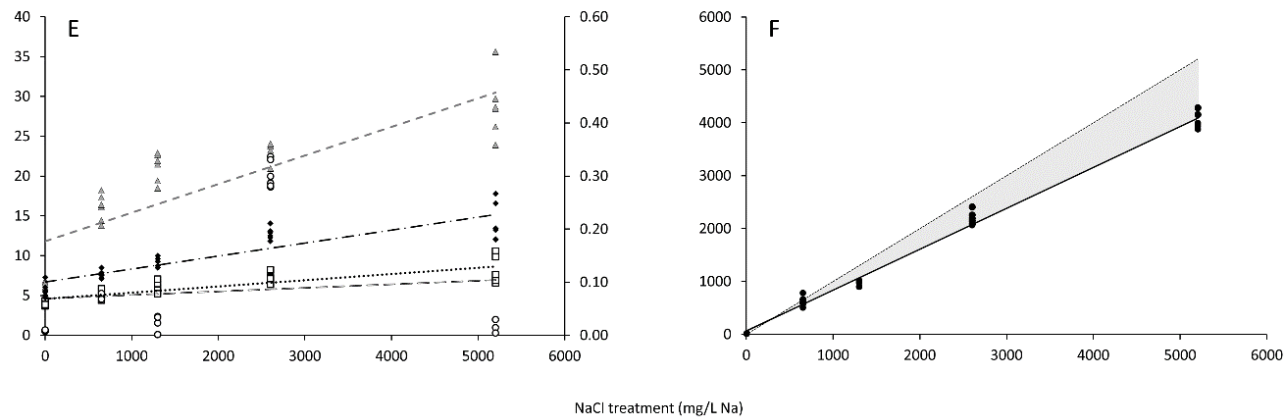
Campus Creek: unrestored



Paint Branch

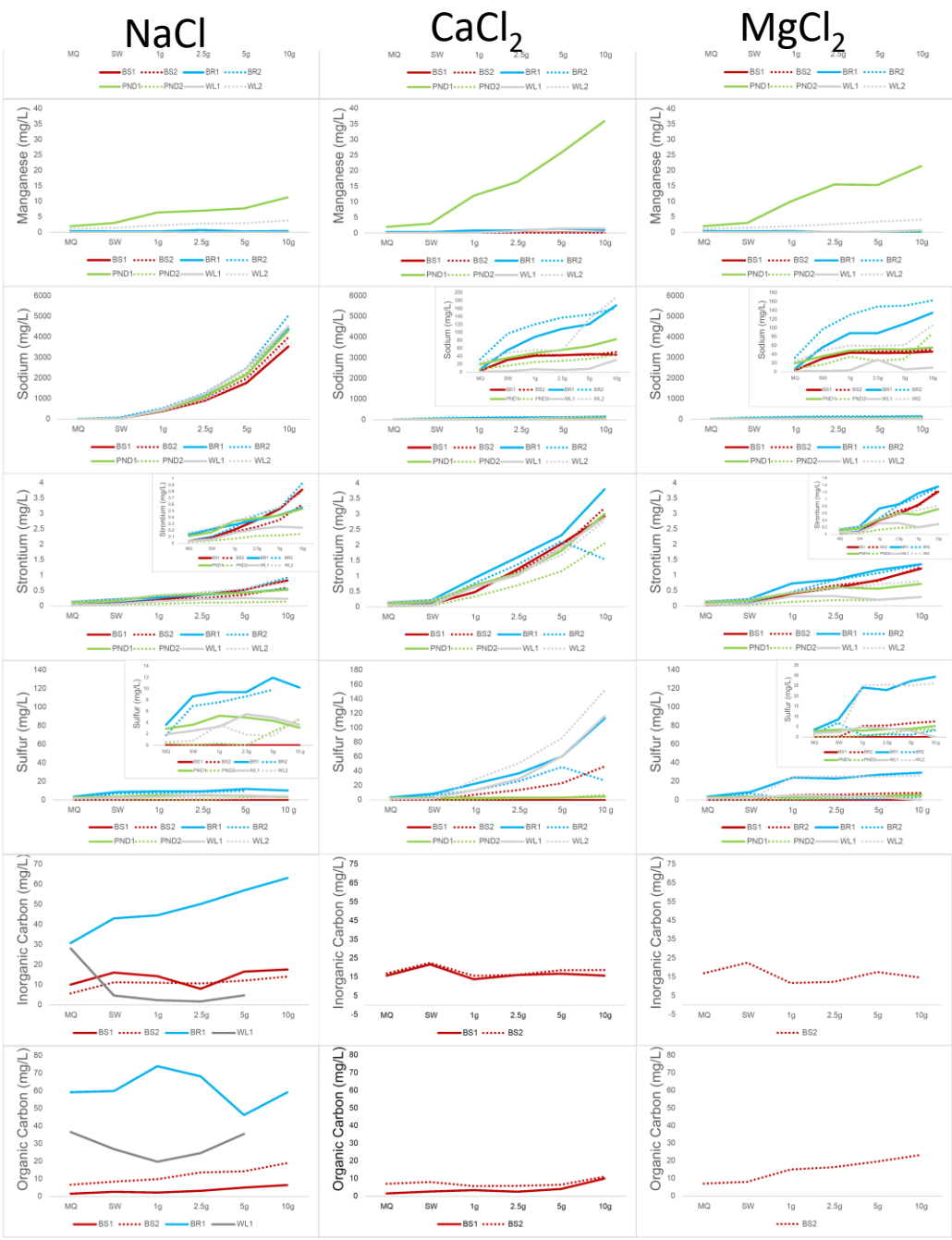
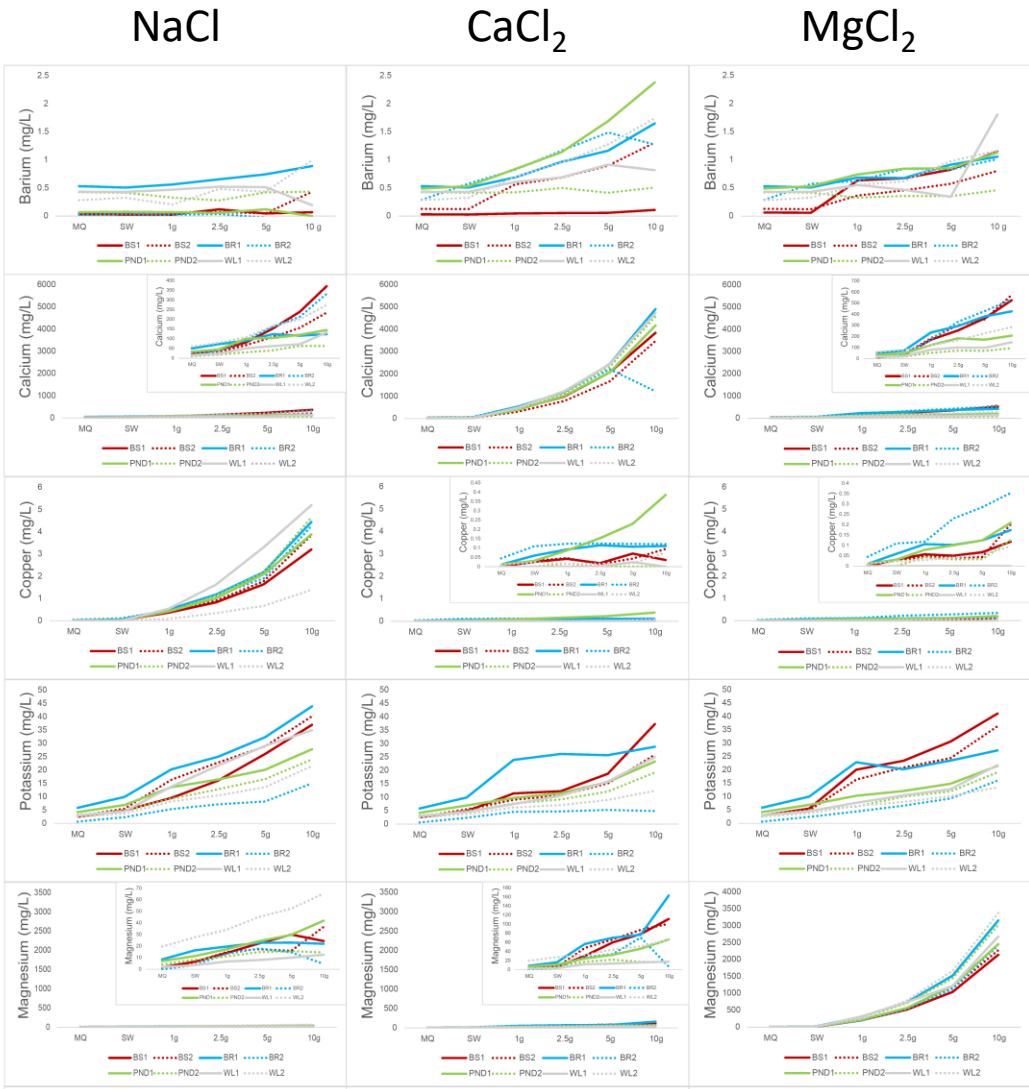


Campus Creek: RSC



Sediments and soils
can retain substantial
amounts of salt ions

Mobilization of metals, nutrients, and other salts

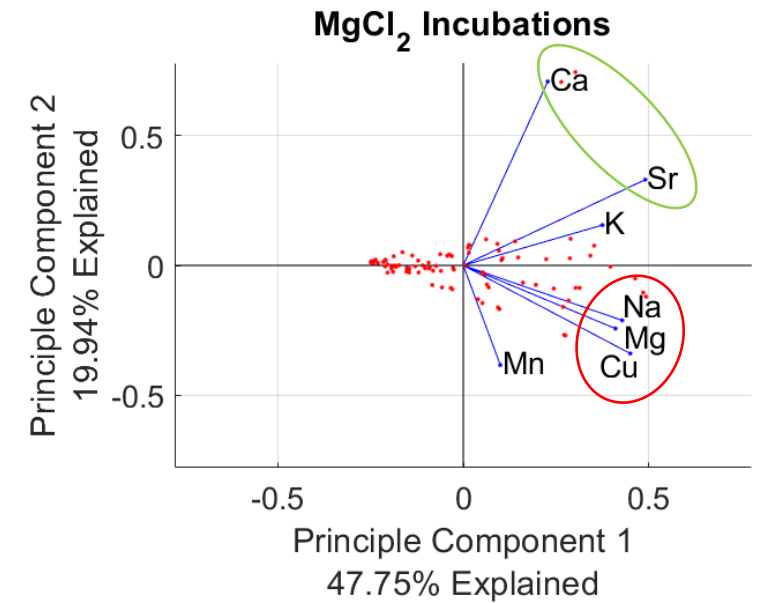
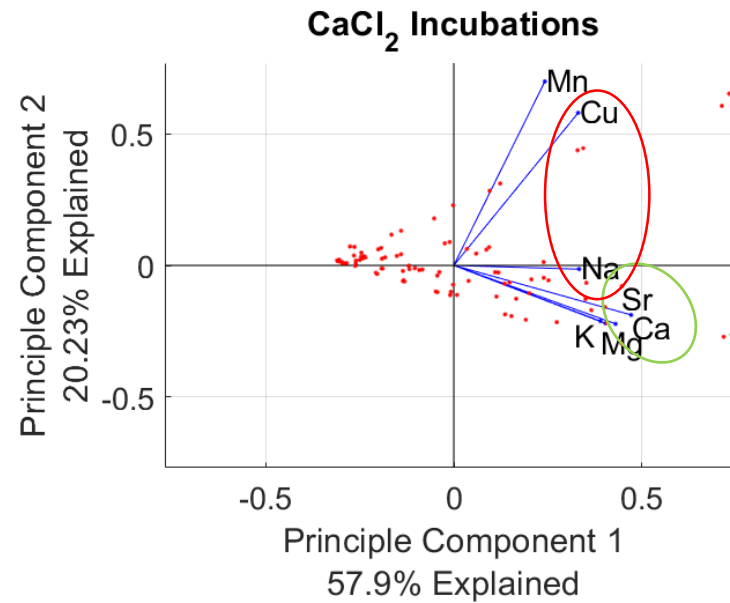
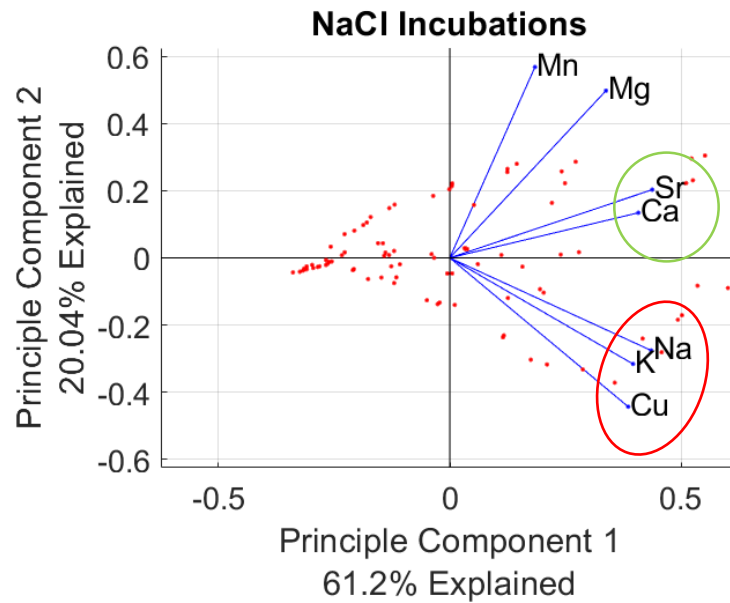


Results: ANOVA Salt type, concentration, and BMP type are significant

- Site type had a significant effect on the concentration of:
 - B, Ba, K, Mn, and Sr
- Salt type was significantly correlated with all major and trace elements save for:
 - B and Fe
- **Salt concentration was statistically correlated with mobilization of ALL major and trace elements**

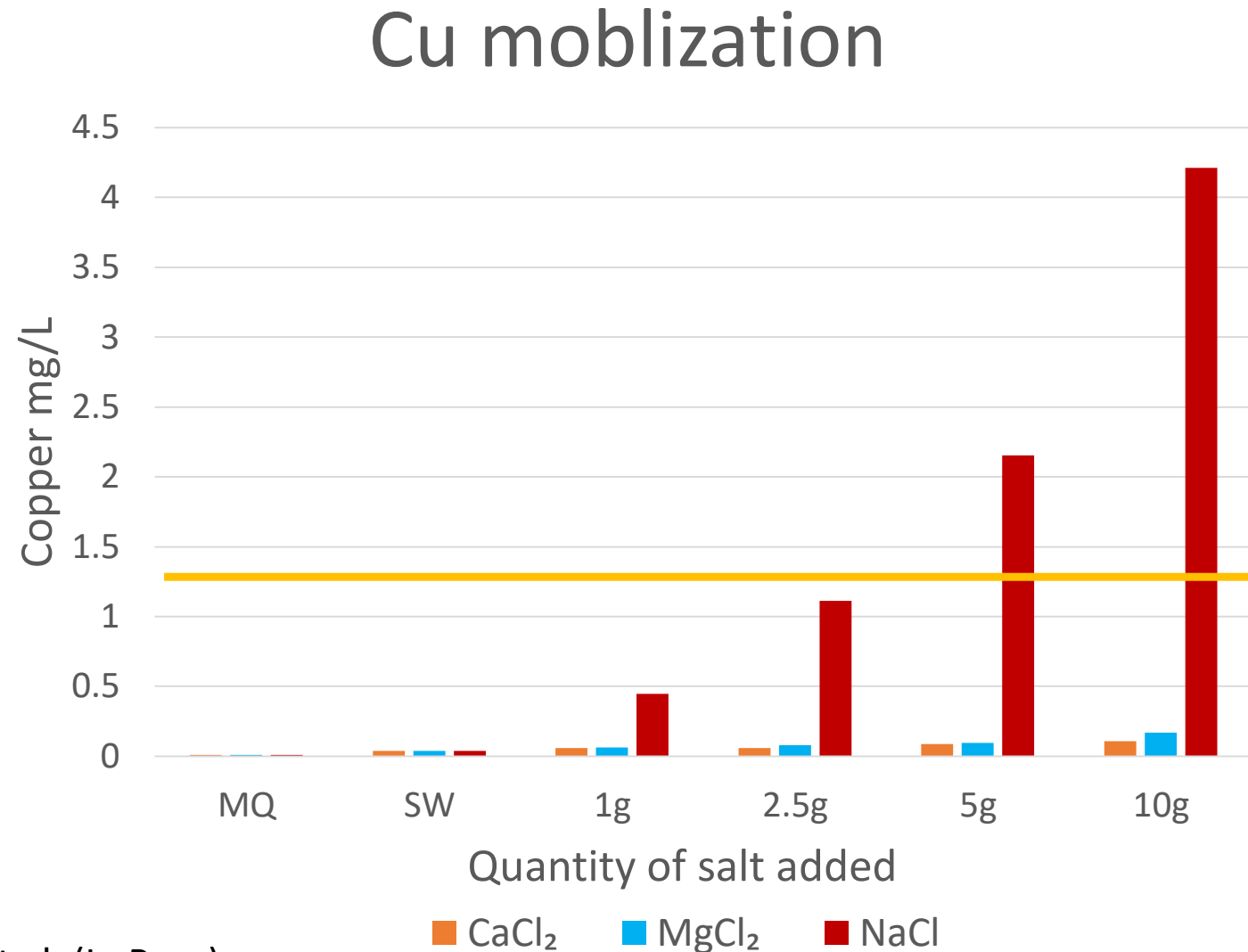
Variable	Site Type			Salt Type			Salt Concentration		
	N	F-ratio	p-value	N	F-ratio	p-value	N	F-ratio	p-value
B	248	10.675	0.000	248	1.989	0.139	248	2.722	0.021
Ba	248	22.458	0.000	248	10.523	0.000	248	26.294	0.000
Ca	248	0.416	0.868	248	69.988	0.000	248	28.744	0.000
Cu	241	0.318	0.927	241	72.073	0.000	241	25.260	0.000
Fe	182	1.784	0.105	182	2.917	0.057	182	2.771	0.020
K	248	41.476	0.000	248	21.122	0.000	248	176.314	0.000
Mg	248	0.420	0.866	248	73.466	0.000	248	23.883	0.000
Mn	245	53.566	0.000	245	3.889	0.022	245	5.324	0.000
Na	248	0.106	0.996	248	77.882	0.000	248	24.069	0.000
Sr	248	6.264	0.000	248	79.050	0.000	248	66.406	0.000
DIC	X	X	X	36	21.301	0.000	36	4.220	0.006
DOC	X	X	X	36	22.903	0.000	36	12.346	0.000
TDN	X	X	X	36	14.548	0.000	36	21.124	0.000

Different Salt Ions Mobilize Different Contaminants



NaCl Mobilizes More Copper: Implications for Aquatic Ecosystems and Home Plumbing

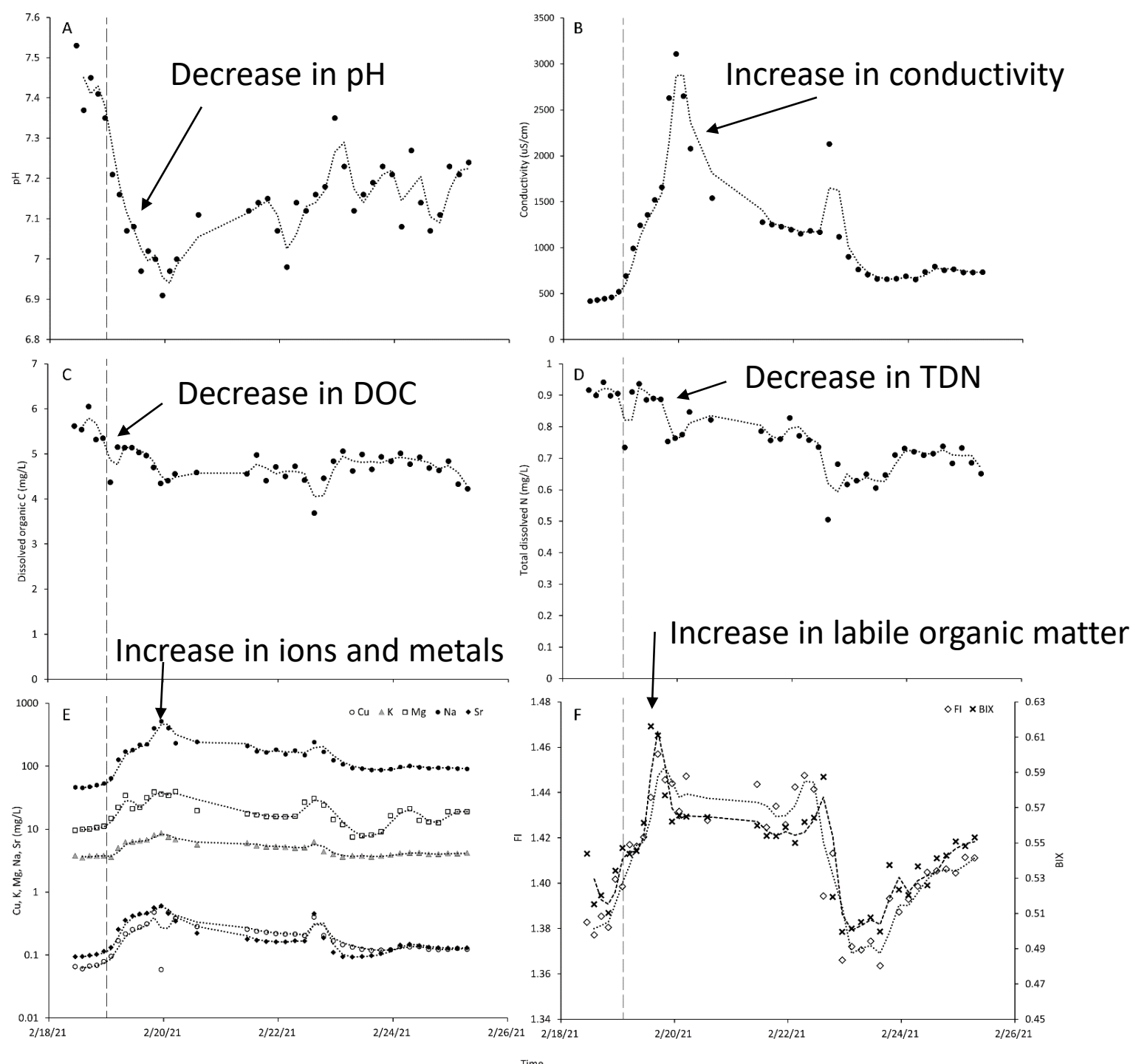
NaCl mobilizes Cu more than an order of magnitude more efficiently than CaCl_2 or MgCl_2



Stream and Stormwater BMP Monitoring



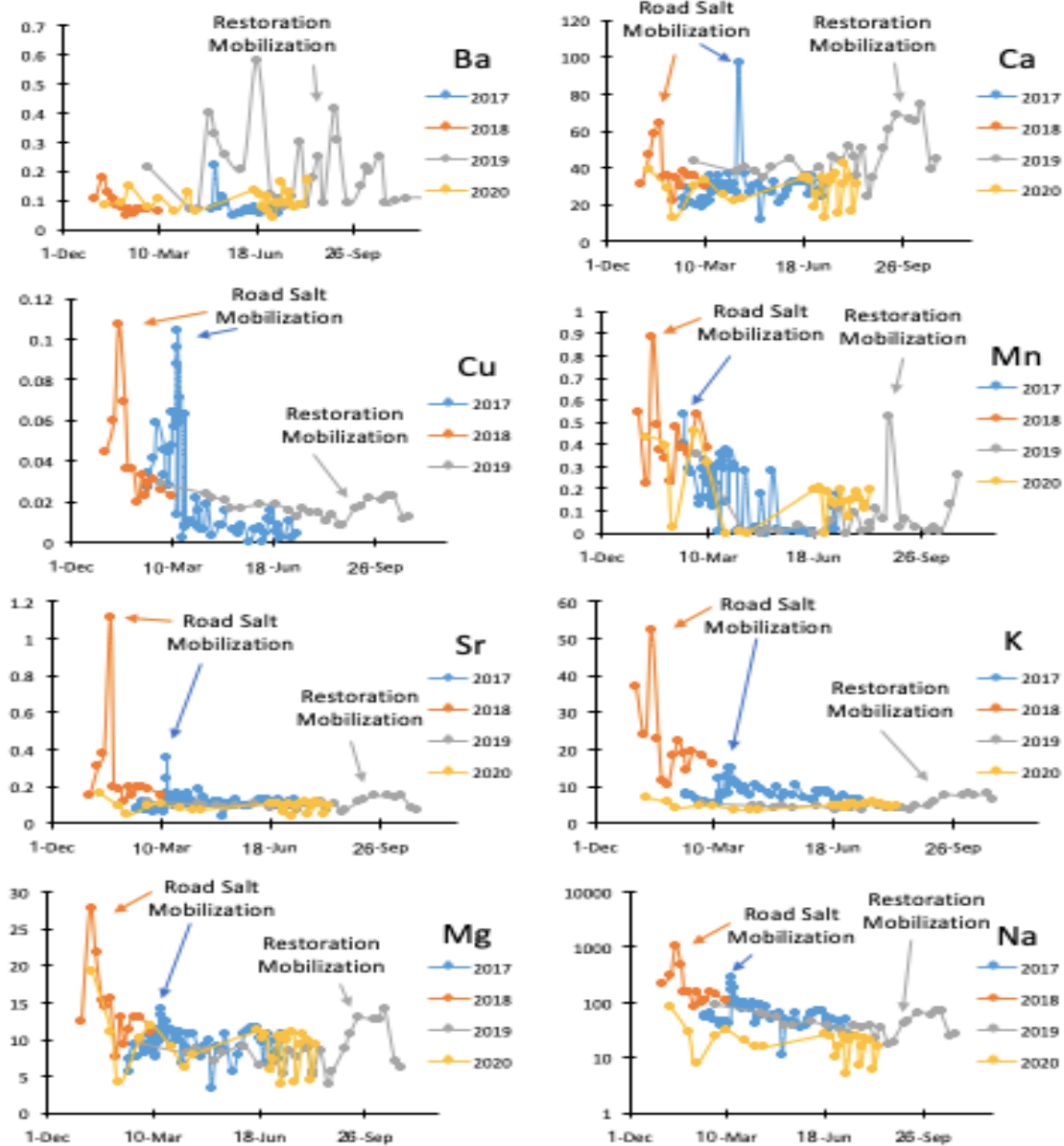
Photos Courtesy: Kelsey Wood



Changes in water quality following road salt events in RSCs

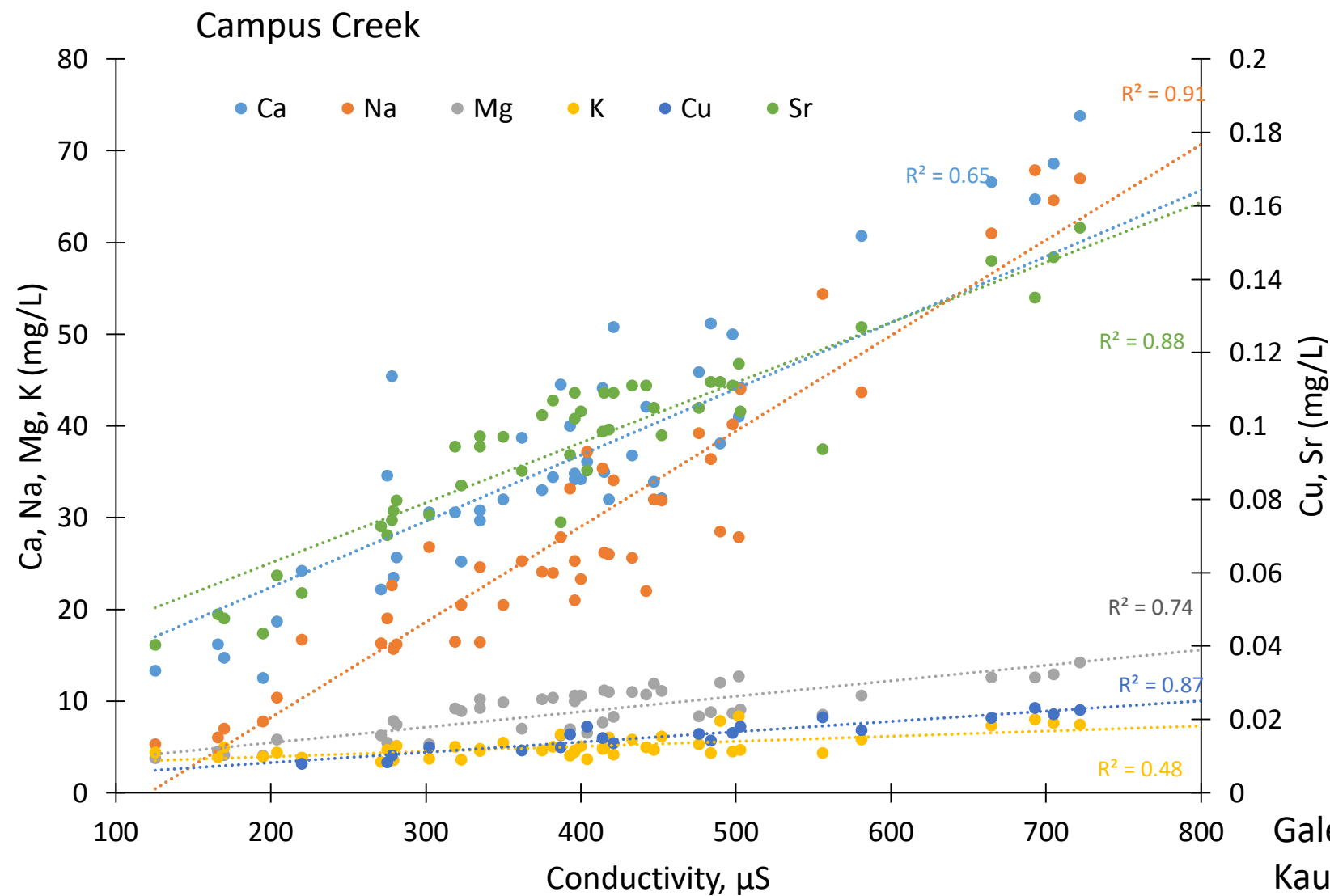
Kaushal et al. (In Press)
Freshwater Science

Mobilizing Chemical Cocktails: Comparison by Year and Season



There can be significant recovery and decreases in ion concentrations over years (>50%) depending on amount of road salt use

New Monitoring Approaches: Specific Conductance as a Proxy for Ions and Metals



Galella et al. (2021),
Kaushal et al. (2021),
Kaushal et al. (2022),
Kaushal et al. (In Press)

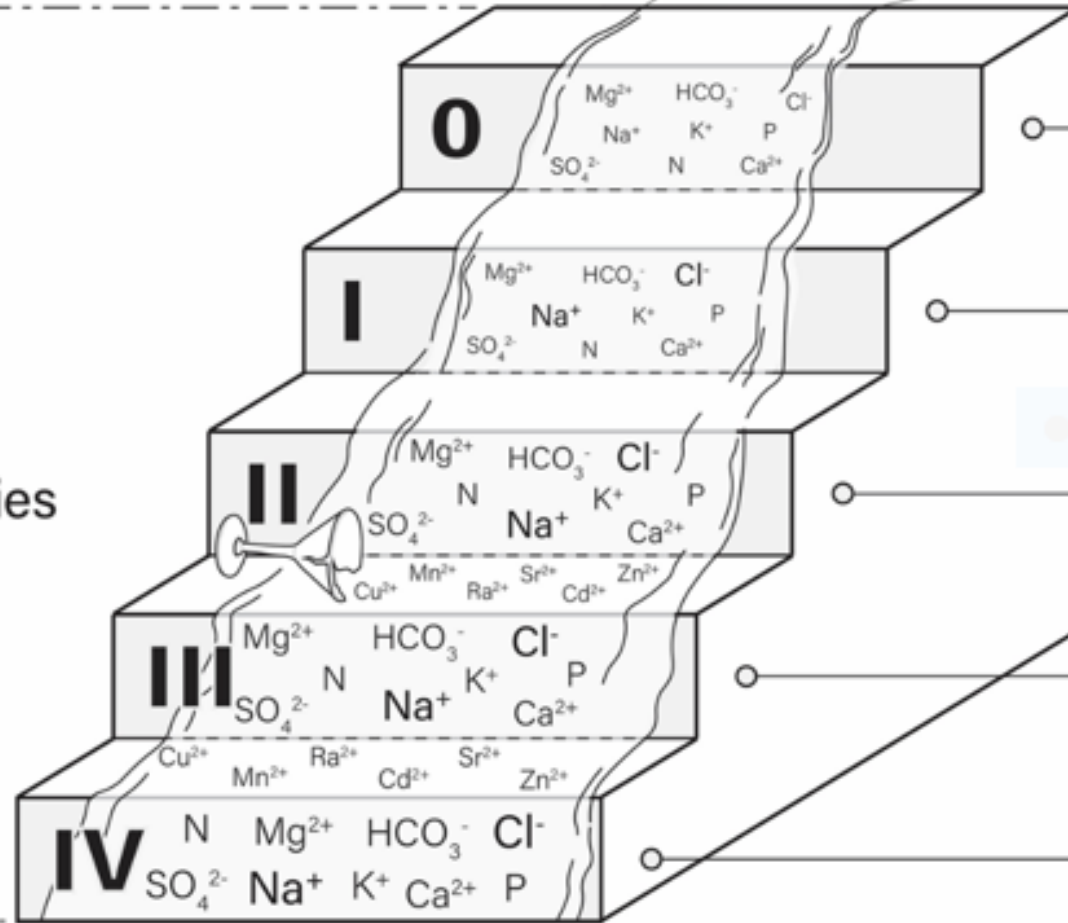
Stages of Freshwater Salinization Syndrome

HIGHEST WATER QUALITY

Driven by
State Factors:

- Climate
- Geology
- Human activities
- Flowpaths
- Time

LOWEST WATER QUALITY



Stage 0. Highest water quality; minimally disturbed.

Stage I. Abnormally elevated concentrations of at least one or more ions across one season.

Stage II. Chronically elevated concentrations of ions across multiple seasons.

Stage III. Formation of harmful chemical cocktails exceeding water quality thresholds.

Stage IV. Systems-level failures in infrastructure and ecosystem functions and services.

Kaushal et al. 2022, *Limnology & Oceanography Letters*

Managing Salinization by the Amounts and Types of Salt Ions along Flowpaths

Summary

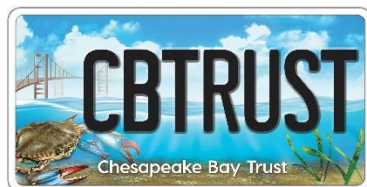
- Significant retention of salt ions in stormwater sediments
- Mobilization of contaminants depends on concentration of salt ions, type of salt, and BMP type
- New practical monitoring approaches using proxies

Management Implications

- Stormwater sediments/soils can have very high potential to enhance ion retention and ion exchange
- Reducing winter NaCl inputs can lead to rapid and year long recovery in some ions
- Water quality monitoring approaches using inexpensive proxies such as specific conductance can help predict concentrations of multiple ions and metals

Acknowledgments

- Thanks to the Maryland Department of Transportation, and Maryland Department of Natural Resources for funding along with all the funding partners below.
- Special thanks also to Karl Berger, Steve Bieber, Paul Mayer, and Stan Grant



The background image shows an outdoor parking lot with asphalt pavement. In the middle ground, there is a large yellow storage container. A red and white vehicle, possibly a utility truck or generator, is parked in front of the container. Several orange traffic cones are placed around the vehicle and container. To the right of the container, there is a large, light-colored bush or shrub. The background is filled with bare trees, suggesting a late autumn or winter setting. The sky is overcast.

Translation Slides

Ken Mack

Senior Water Quality Specialist

Montgomery County Department of Environmental Protection

Good News Bad News

Good News

- BMPs and Stream Restoration are sequestering deicing salts

Bad News

- Sequestered deicers are exchanged for other ions (Including environmentally harmful ions)



Good News Bad News

Good News

- Reducing road salt use results in immediate reduction in stream salinization and reduces ion exchange

Bad News

- NaCl results in a likely more harmful chemical cocktail (especially in bioretention)



Good News Bad News

Good News

- Conductivity monitoring can be a good proxy for several ions (not just Na or Cl)

Bad News

- Exchanged ions result in a broad chemical cocktail possibly reducing biological potential



Additional Management Considerations

- Emphasizes the need to reduce the use of all salts (especially NaCl)
- Many second order effects of deicers (acidification, increase in labile organic matter, reduction in DOC). Potential impacts to:
 - Drinking water
 - Stream biota
 - Infrastructure
- Salt type has a major influence on released chemical cocktail
 - Consider using different salts near bioretention
 - Reduce NaCl where copper export is possible
- BMP type and age influences released chemical cocktail
- Stream restoration results in a pulse of nutrients/ions

